

FITC Substitution Doubles the Dynamic Range of a Fluorescein-tagged Oligoribonucleotide pH Sensor Kaveen Harohalli (C 2025), Madison R. Herling, and Ivan J. Dmochowski, Ph.D. Department of Chemistry, School of Arts and Sciences, University of Pennsylvania

Introduction

- Within the physiologic pH range of 4.0-7.5, the pH sensitivity of the xanthene dye fluorescein originates primarily from the equilibrium between the monoanionic and dianionic species.
- Since the pK_a corresponding to this equilibrium sits well within the range of biologically relevant pH values, fluorescein-biomolecule conjugates have found use as pH sensors.
- The pK_a itself is sensitive to the electrostatic environment around the dye and can be perturbed by judicious alteration of the biomolecule.



| Anal. Chem. **2007**, 79, 6775-6782.

Background and Motivation

- We have found that FAM (6-carboxyfluorescein) -labeled pH probes are adequately sensitive in buffer but not in cells.
- The thiourea linker arising from FITC (fluorescein isothiocyanate) labeling is a known fluorescein quencher in peptides. We hypothesized that this quenching could work synergistically with the intrinsic pH sensitivity of fluorescein to increase the sensitivity.
- If the thiourea can perform this function in a labeled oligo, this heightened sensitivity may allow us to deploy a variety of oligobased highly sensitive pH sensors in live cells.



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Characterization and Evaluation



5'-FITC-(2'-OMe)U₅-3' is more pH-sensitive than the FAM analogue.

Future Directions

- Improve the synthesis of FITC-labeled oligos to increase recovery.
- Confirm the mechanism of fluorescence quenching and determine if the observed effect is as pH-dependent as it appears to be.
- Develop and deploy a FITC-labeled oligo to monitor pH in live cells.

References

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Johnson, O. T.; Kaur, T.; Garner, A. T. A conditionally fluorescent peptide reporter of secondary structure modulation. ChemBioChem 2019, 20, 40-45.