

# Air Pollutant Chemistry at the Air-Lung Interface: An Interdisciplinary Study

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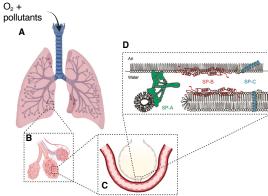
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## **Objectives**

This thesis built on biological approaches to modeling pulmonary surfactant by integrating atmospheric chemistry approaches to simulating the chemical reactivity of NO<sub>2</sub> at the air-lung interface. We aimed to answer the following questions: a) what is the orientation of inhaled atmospheric NO<sub>2</sub> with respect to biological molecules in surfactant, and b) based on this orientation, what chemical transformations does NO<sub>2</sub> undergo in this environment?

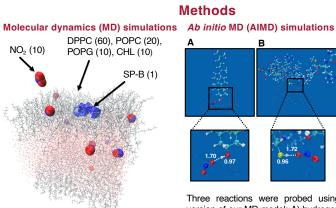
## Background

- Nitrogen dioxide (NO<sub>2</sub>) is a gaseous air pollutant and reactive nitrogen species. Exposure has been associated with respiratory disease, cardiovascular disease, and cancer, but the mechanism of action underlying NO<sub>2</sub> toxicology remains uncertain.
- · Pulmonary surfactant (PS), the top layer of the lung lining fluid, is a complex mixture of lipids and proteins.1 Situated at the air-lung interface, PS forms the first point of contact for inhaled atmospheric pollutants and therefore has been the subject of several investigations on the effect of foreign particles in the lung.



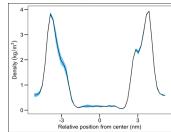
Pathway of air exchange in the lungs. Panel D is reprinted from Parra & Pérez-Gil (2015).1

- The effects of NO<sub>2</sub> on PS have not been experimentally tested, but exposing PS to reactive oxygen species has been shown to result in chemical modifications to lipids and proteins.
- · These PS lipids and proteins have several functional groups which have been shown to react with NO<sub>2</sub> in other environments.
  - NO<sub>2</sub> can oxidize the thiol functional group in cysteine (Cys),<sup>2</sup> and although the hydroxyl group on cholesterol (CHL) is less acidic, a similar oxidation pathway has been shown computationally.3
- · Furthermore, both thiol and hydroxyl oxidation reactions have been observed to accelerate at air-water interfaces such as PS. Water has been shown to act as a catalyst at these aqueous interfaces. an effect known as "on-water catalysis."4
- · To our knowledge, only one study has investigated the effect of atmospheric pollutants on PS using computational methods, and the methodology did not allow consideration of chemical reactivity.5

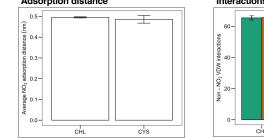


PS was modeled using a bilayer separated by a water slab in a 6.5x6.5x8.5 nm<sup>3</sup> box. Simulations were run in GROMACS for 200 ns.

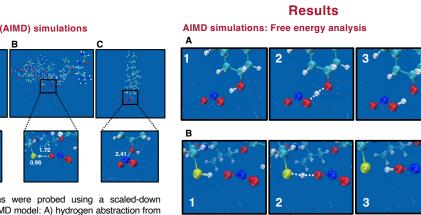
#### MD simulations: Adsorption analysis NO<sub>2</sub> mass density profile



#### Adsorption distance



NO<sub>2</sub> molecules adopted a symmetric bimodal distribution during the simulations, preferentially locating in and around the lipid layers. Adsorption of NO2 on CHL and Cys was equally likely and equally strong as measured by distance and number of non-NO2 interactions broken.



С n -1.0 -0.5 0.0 0.5 -0.5 0.0 0.5 Collective variable (Å) Collective variable (Å)

Hydrogen abstraction by NO<sub>2</sub> from both CHL hydroxyl (A) and Cys thiol (B) was observed, yielding HONO(+) and alkoxy or thiolate anions, respectively. Only the Cys thiol pathway was found to be energetically spontaneous (C. D).

## Conclusions

- · Our simulations established the preferential distribution of NO<sub>2</sub> in the lipid laver of PS near functional groups such as CHL and Cvs. suggesting the possibility of chemical reactions occurring.
- · Given the comparable strength of adsorption for both functional groups, our AIMD simulations suggest that hydrogen abstraction by Cvs thiol in PS is a more favorable process.
- Future work should explore the health implications of HONO(+) and of the functional group modifications observed to yield additional insight into the molecular mechanism by which NO2 impacts health.

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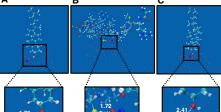
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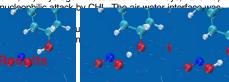




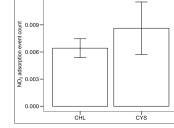




Three reactions were probed using a scaled-down version of our MD model: A) hydrogen abstraction from CHL, B) hydrogen abstraction from Cvs. and C)



0.012



### Interactions with non-NO<sub>2</sub> atoms

